

# Ferroelectric and piezoelectric properties of $[(\text{Bi}_{0.98}\text{La}_{0.02}\text{Na}_{1-x}\text{Li}_x)_{0.5}]_{0.94}\text{Ba}_{0.06}\text{TiO}_3$ lead-free ceramics

Dunmin Lin · K. W. Kwok

Received: 29 March 2009 / Accepted: 15 July 2009 / Published online: 29 July 2009  
© Springer Science+Business Media, LLC 2009

**Abstract** Lead-free ceramics  $[(\text{Bi}_{0.98}\text{La}_{0.02}\text{Na}_{1-x}\text{Li}_x)_{0.5}]_{0.94}\text{Ba}_{0.06}\text{TiO}_3$  have been prepared by an ordinary sintering technique and their ferroelectric and piezoelectric properties have been studied. The results of X-ray diffraction reveal that  $\text{Li}^+$ ,  $\text{Ba}^{2+}$ , and  $\text{La}^{3+}$  diffuse into the  $\text{Bi}_{0.5}\text{Na}_{0.5}\text{TiO}_3$  lattices to form a new solid solution with a pure perovskite structure. The partial substitution of  $\text{Li}^+$  lowers the coercive field  $E_c$  and improves the remanent polarization  $P_r$ . Because of the larger  $P_r$  and lower  $E_c$ , the ceramic with  $x = 0.10$  exhibits optimum piezoelectric properties:  $d_{33} = 212$  pC/N and  $k_p = 36.1\%$ . The partial substitution of  $\text{Li}^+$  for  $\text{Na}^+$  shifts the depolarization temperature  $T_d$  toward low temperature. The ceramics exhibit relaxor characteristic, which is probably resulted from the cation disordering in the 12-fold coordination sites. The temperature dependences of the ferroelectric and dielectric properties suggest that the ceramics contain both the polar and non-polar regions near/above  $T_d$ , which cause the polarization hysteresis loop become deformed and the ceramics become depolarized.

## Introduction

Lead-based piezoelectric ceramics with perovskite structure based on lead zirconate titanate (PZT) and PZT-based multi-component systems are widely used for piezoelectric

actuators, sensors, transducers as well as microelectronic devices because of their excellent piezoelectric properties. However, the use of lead-based materials has caused serious lead pollution and environmental problems because of the high toxicity of lead oxide. Therefore, it is necessary to develop environment-friendly lead-free ferroelectric and piezoelectric ceramics.

$\text{Bi}_{0.5}\text{Na}_{0.5}\text{TiO}_3$  (BNT) ceramic is a perovskite-structured ferroelectric with rhombohedral symmetry. Because of its strong ferroelectricity ( $P_r = 38$   $\mu\text{C}/\text{cm}^2$ ) [1], the BNT ceramic has been considered as one of the promising candidates for lead-free ceramics. However, it also has a high coercive field ( $E_c = 7.3$  kV/mm) [1], resulting in the difficulty in the poling of the ceramic. Therefore, the pure BNT ceramic usually exhibits relatively weak piezoelectric properties ( $d_{33} = 58$  pC/N) [2]. A number of studies have been carried out to improve the poling process and enhance the piezoelectric properties of the ceramics; these include the formation of solid solutions of BNT with other ABO<sub>3</sub>-type ferroelectrics or non-ferroelectrics, e.g., BNT–BaTiO<sub>3</sub> [1], BNT–Bi<sub>0.5</sub>K<sub>0.5</sub>TiO<sub>3</sub> [3], BNT–BiAlO<sub>3</sub> [4], BNT–Bi<sub>0.5</sub>K<sub>0.5</sub>TiO<sub>3</sub>–KNbO<sub>3</sub> [5], BNT–KNbO<sub>3</sub> [6], BNT–SrTiO<sub>3</sub> [7], BNT–Bi<sub>0.5</sub>K<sub>0.5</sub>TiO<sub>3</sub>–BaTiO<sub>3</sub> [8, 9],  $[\text{Bi}_{0.5}(\text{Na}_{0.7}\text{K}_{0.25}\text{Li}_{0.05})_{0.5}]\text{TiO}_3$ –Ba(Ti<sub>0.95</sub>Zr<sub>0.05</sub>)O<sub>3</sub> [10], BNT–K<sub>0.5</sub>Na<sub>0.5</sub>NbO<sub>3</sub> [11], BNT–BNT–Bi<sub>0.5</sub>K<sub>0.5</sub>TiO<sub>3</sub>–0.03K<sub>0.5</sub>Na<sub>0.5</sub>NbO<sub>3</sub> [12], and BNT–Bi<sub>0.5</sub>K<sub>0.5</sub>TiO<sub>3</sub>–BiFeO<sub>3</sub> [13], the substitutions of analogous ions for the A-site ( $\text{Bi}_{0.5}\text{Na}_{0.5}$ )<sup>+</sup> or B-site Ti<sup>4+</sup> ions, e.g.,  $(\text{Bi}_{1/2}\text{Na}_{1/2})\text{Ti}_{1-x}(\text{Ni}_{1/3}\text{Nb}_{2/3})_x\text{O}_3$  [14] and  $(\text{Bi}_{0.5}\text{Na}_{0.5})_{0.94}\text{Ba}_{0.06}\text{Zr}_y\text{Ti}_{1-y}\text{O}_3$  [15], and the doping of metal oxides, e.g., Nd<sub>2</sub>O<sub>3</sub>-doped 0.82BNT–0.18Bi<sub>0.5</sub>K<sub>0.5</sub>TiO<sub>3</sub> [16], Ta-doped 0.94BNT–0.06BaTiO<sub>3</sub> [17], and CeO<sub>2</sub>-doped Bi<sub>0.5</sub>Na<sub>0.44</sub>K<sub>0.06</sub>TiO<sub>3</sub> [18]. It has been noted that as a classical BNT-based system, BNT–BaTiO<sub>3</sub> (BNT–BT) ceramics have been reported frequently [1, 17, 19, 20]. For BNT–BT ceramics, the optimum piezoelectric

D. Lin (✉)  
College of Chemistry and Materials Science, Sichuan Normal University, Chengdu 610066, People's Republic of China  
e-mail: ddmd222@yahoo.com.cn

K. W. Kwok  
Department of Applied Physics and Materials Research Centre, The Hong Kong Polytechnic University, Kowloon, Hong Kong, People's Republic of China

properties are achieved at 6 mol.% BaTiO<sub>3</sub> because of the existence of morphotropic phase boundary (MPB) but the maximum value of piezoelectric coefficient  $d_{33}$  (125 pC/N) is not high enough [1]. Our previous work has also shown that the substitution of a small amount of La<sup>3+</sup> (2–4 mol.%) for Bi<sup>3+</sup> in the 0.94BNT–0.06BaTiO<sub>3</sub> ceramics can effectively enhance the piezoelectric properties [21]. In the present work, a new BNT-based multi-component solid solution, [(Bi<sub>0.98</sub>La<sub>0.02</sub>Na<sub>1-x</sub>Li<sub>x</sub>)<sub>0.5</sub>]<sub>0.94</sub>Ba<sub>0.06</sub>TiO<sub>3</sub>, was developed by the partial substitutions of 2.5–12.5 mol.% Li<sup>+</sup> for Na<sup>+</sup> and 2 mol.% La<sup>3+</sup> for Bi<sup>3+</sup> in the A-sites of 0.94BNT–0.06BaTiO<sub>3</sub> ceramics and prepared by an ordinary sintering method, and their microstructure, ferroelectric, and piezoelectric properties were investigated.

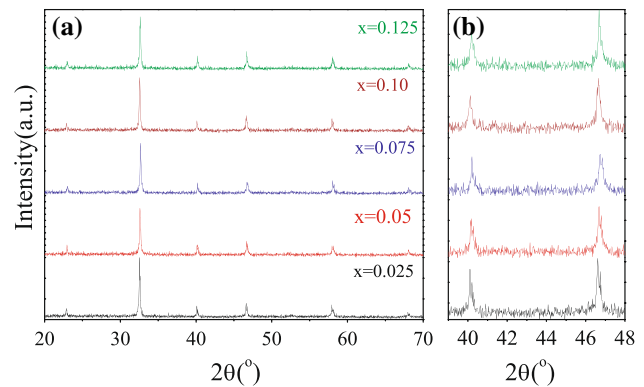
## Experimental

[(Bi<sub>0.98</sub>La<sub>0.02</sub>Na<sub>1-x</sub>Li<sub>x</sub>)<sub>0.5</sub>]<sub>0.94</sub>Ba<sub>0.06</sub>TiO<sub>3</sub> ceramics were prepared by a conventional ceramic fabrication technique using analytical-grade metal oxides or carbonate powders as raw materials: Bi<sub>2</sub>O<sub>3</sub> (99%), Na<sub>2</sub>CO<sub>3</sub> (99%), Li<sub>2</sub>CO<sub>3</sub> (97%), BaCO<sub>3</sub> (99%), La<sub>2</sub>O<sub>3</sub> (99%), and TiO<sub>2</sub> (99.5%). The powders in the stoichiometric ratio of the compositions were mixed thoroughly in ethanol using zirconia balls for 8 h, and then dried and calcined at 850 °C for 2 h. After the calcination, the mixture was ball milled again for 8 h and mixed thoroughly with a PVA binder solution, and then pressed into disk samples. The disk samples were sintered at 1,100 °C for 2 h in air. For electrical measurements, silver electrodes were fired on the top and bottom surfaces of the samples at 730 °C for 15 min. The samples were poled at room temperature under a dc field of 5–6 kV/mm in a silicon oil bath for 30 min.

The crystalline structure of the sintered samples was examined using X-ray diffraction (XRD) analysis with CuK<sub>α</sub> radiation (DX-1000). The microstructure was observed using a scanning electron microscope (JEOL JSM-5900LV). The relative permittivity  $\epsilon_r$  and loss tangent  $\tan\delta$  of the ceramics at 1, 10, and 100 kHz were measured as functions of temperature using an impedance analyzer (Agilent 4192A). A conventional Sawyer-Tower circuit was used to measure the polarization hysteresis ( $P$ – $E$ ) loop at 50 Hz. The planar electromechanical coupling factor  $k_p$  and mechanical quality factor  $Q_m$  were determined by the resonance method according to the IEEE Standard using an impedance analyzer (Agilent 4294A). The piezoelectric coefficient  $d_{33}$  was measured using a piezo- $d_{33}$  meter (ZJ-3A, China).

## Results and discussion

The XRD patterns of the [(Bi<sub>0.98</sub>La<sub>0.02</sub>Na<sub>1-x</sub>Li<sub>x</sub>)<sub>0.5</sub>]<sub>0.94</sub>Ba<sub>0.06</sub>TiO<sub>3</sub> ceramics are shown in Fig. 1. All the ceramics

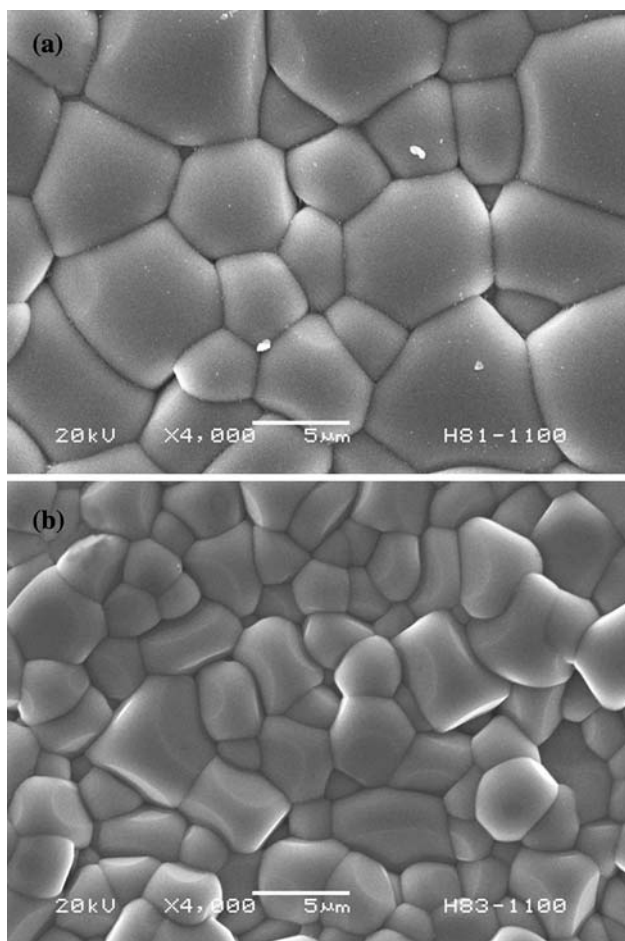


**Fig. 1** XRD patterns of the [(Bi<sub>0.98</sub>La<sub>0.02</sub>Na<sub>1-x</sub>Li<sub>x</sub>)<sub>0.5</sub>]<sub>0.94</sub>Ba<sub>0.06</sub>TiO<sub>3</sub> ceramics

possess a pure perovskite structure, suggesting that Li<sup>+</sup>, Ba<sup>2+</sup>, and La<sup>3+</sup> have diffused into the BNT lattices to form a new homogeneous solid solution (Fig. 1a). Similar to the 0.94BNT–0.06BaTiO<sub>3</sub> ceramic [1], all the [(Bi<sub>0.98</sub>La<sub>0.02</sub>Na<sub>1-x</sub>Li<sub>x</sub>)<sub>0.5</sub>]<sub>0.94</sub>Ba<sub>0.06</sub>TiO<sub>3</sub> ceramics reside within the MPB. This is evidenced by the splitting of the (003)/(021) characteristic peaks between 39° and 41° and the splitting of the (002)/(200) characteristic peaks between 46° and 48° as observed in the [(Bi<sub>0.98</sub>La<sub>0.02</sub>Na<sub>1-x</sub>Li<sub>x</sub>)<sub>0.5</sub>]<sub>0.94</sub>Ba<sub>0.06</sub>TiO<sub>3</sub> ceramics (Fig. 1b), indicating that the substitutions of Li<sup>+</sup> for Na<sup>+</sup> and 2 mol.% La<sup>3+</sup> for Bi<sup>3+</sup> do not cause any significant change to the crystalline structure.

The SEM micrographs of the [(Bi<sub>0.98</sub>La<sub>0.02</sub>Na<sub>1-x</sub>Li<sub>x</sub>)<sub>0.5</sub>]<sub>0.94</sub>Ba<sub>0.06</sub>TiO<sub>3</sub> ceramics with  $x = 0.05$  and  $0.10$  are shown in Fig. 2. Both the ceramics are well-sintered at 1,100 °C for 2 h, are dense and pore-free, having a relative density (measured by the Archimedes method) larger than 97%. It can be seen that the substitution of Li is effective in suppressing the grain growth. For the ceramic with  $x = 0.05$ , the average grain size is about 5.7 μm (Fig. 2a). As  $x$  increases to 0.10, the grain size decreases significantly to ~3.0 μm (Fig. 1b). Similar results have been observed for the BNT–Bi<sub>0.5</sub>K<sub>0.5</sub>TiO<sub>3</sub> ceramics [22], for which the grain size was reduced significantly after the addition of K<sup>+</sup>. It has been noted that, as compared to the Li-free BNT-based ceramics, the [(Bi<sub>0.98</sub>La<sub>0.02</sub>Na<sub>1-x</sub>Li<sub>x</sub>)<sub>0.5</sub>]<sub>0.94</sub>Ba<sub>0.06</sub>TiO<sub>3</sub> ceramics can be well-sintered at a lower temperature (1,100 vs. 1,150–1,250 °C [1–10]). This may be attributed to the formation of the liquid phase arisen from the low melting temperature of the Li-containing compounds.

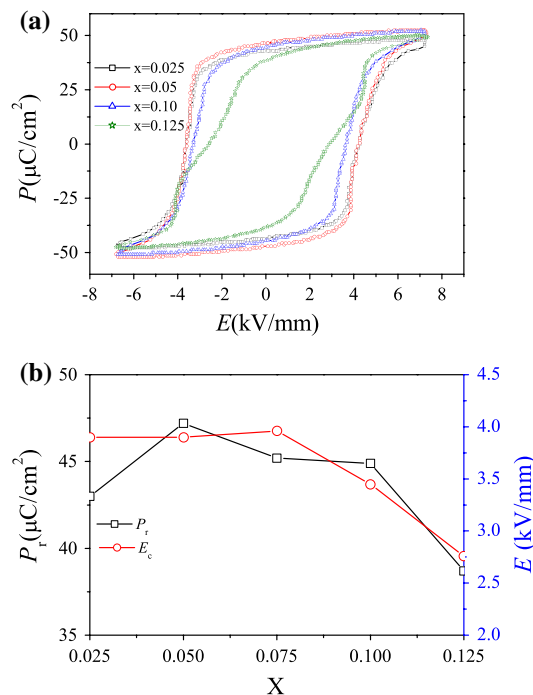
The  $P$ – $E$  loops measured under an electric field of 7 kV/mm at room temperature for the [(Bi<sub>0.98</sub>La<sub>0.02</sub>Na<sub>1-x</sub>Li<sub>x</sub>)<sub>0.5</sub>]<sub>0.94</sub>Ba<sub>0.06</sub>TiO<sub>3</sub> ceramics with  $x = 0.025$ , 0.05, 0.10, and 0.125 are shown in Fig. 3a, while the compositional dependences of the remanent polarization  $P_r$  and coercive field  $E_c$  are shown in Fig. 3b. All the ceramics exhibit a



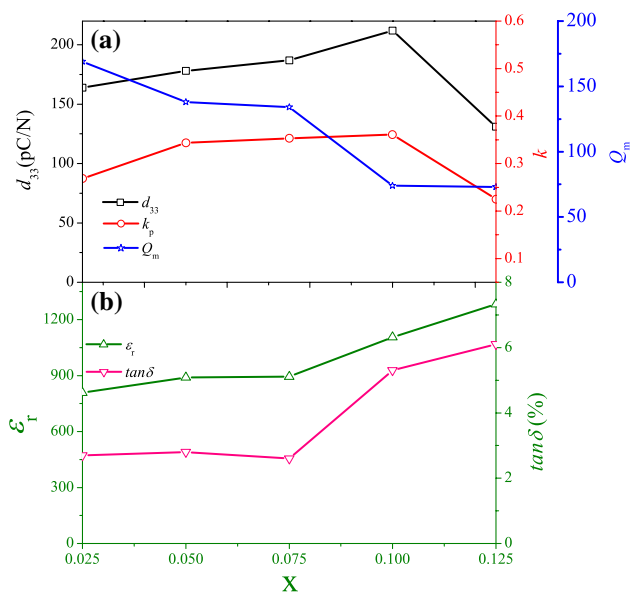
**Fig. 2** SEM micrographs of the  $[(\text{Bi}_{0.98}\text{La}_{0.02}\text{Na}_{1-x}\text{Li}_x)_{0.5}]_{0.94}\text{Ba}_{0.06}\text{TiO}_3$  ceramics with **a**  $x = 0.05$  sintered at  $1,100\text{ }^\circ\text{C}$  for 2 h; **b**  $x = 0.10$  sintered at  $1,100\text{ }^\circ\text{C}$  for 2 h

typical and saturated  $P$ - $E$  loop. As shown in Fig. 3a, the ceramic with  $x = 0.025$  exhibits well-saturated and square-like  $P$ - $E$  loops with  $P_r$  and  $E_c$  of  $43.0\text{ }\mu\text{C}/\text{cm}^2$  and  $3.90\text{ kV}/\text{mm}$ . As  $x$  increases from 0.025 to 0.10, the  $E_c$  is lowered but simultaneously the  $P_r$  is increased. However, as  $x$  further increases to 0.125, the  $P$ - $E$  loop becomes slightly flattened and slanted, giving a much smaller  $P_r$ . As shown in Fig. 3b, the observed  $P_r$  increases with increasing  $x$  and then decreases, giving a maximum value of  $47.2\text{ }\mu\text{C}/\text{cm}^2$  at  $x = 0.05$ , while the observed  $E_c$  decreases from 3.90 to  $2.76\text{ kV}/\text{mm}$  as  $x$  increases from 0.025 to 0.125. As compared to the pure BNT ceramic ( $P_r = 38\text{ }\mu\text{C}/\text{cm}^2$  and  $E_c = 7.3\text{ kV}/\text{mm}$ ), the present ceramics possess the much larger  $P_r$  and lower  $E_c$ . The large remanent polarization favors the piezoelectricity, while the low coercive field should facilitate the poling process.

The variations of  $d_{33}$ ,  $k_p$ ,  $Q_m$ ,  $\epsilon_r$ , and  $\tan\delta$  with  $x$  for the  $[(\text{Bi}_{0.98}\text{La}_{0.02}\text{Na}_{1-x}\text{Li}_x)_{0.5}]_{0.94}\text{Ba}_{0.06}\text{TiO}_3$  ceramics are shown in Fig. 4. As shown in Fig. 4a, the observed  $d_{33}$  increases with increasing  $x$  and then decreases, giving the



**Fig. 3** **a**  $P$ - $E$  loops for the  $[(\text{Bi}_{0.98}\text{La}_{0.02}\text{Na}_{1-x}\text{Li}_x)_{0.5}]_{0.94}\text{Ba}_{0.06}\text{TiO}_3$  ceramics with  $x = 0.025, 0.05, 0.10,$  and  $0.125$  at room temperature; **b** Variations of  $P_r$  and  $E_c$  of the  $[(\text{Bi}_{0.98}\text{La}_{0.02}\text{Na}_{1-x}\text{Li}_x)_{0.5}]_{0.94}\text{Ba}_{0.06}\text{TiO}_3$  ceramics with  $x$



**Fig. 4** Compositional dependences of  $d_{33}$ ,  $k_p$ ,  $Q_m$ ,  $\epsilon_r$ , and  $\tan\delta$  for the  $[(\text{Bi}_{0.98}\text{La}_{0.02}\text{Na}_{1-x}\text{Li}_x)_{0.5}]_{0.94}\text{Ba}_{0.06}\text{TiO}_3$  ceramics

maximum values of  $212\text{ pC}/\text{N}$  at  $x = 0.10$ . The observed  $k_p$  exhibits a similar variation with  $x$  and has an optimum value of  $36.1\%$  at  $x = 0.10$  (Fig. 4a). From Fig. 4b, at  $x \leq 0.075$ , the observed  $\epsilon_r$  and  $\tan\delta$  have weak dependences on  $x$  but when  $x > 0.075$ , the  $\epsilon_r$  and  $\tan\delta$  increase

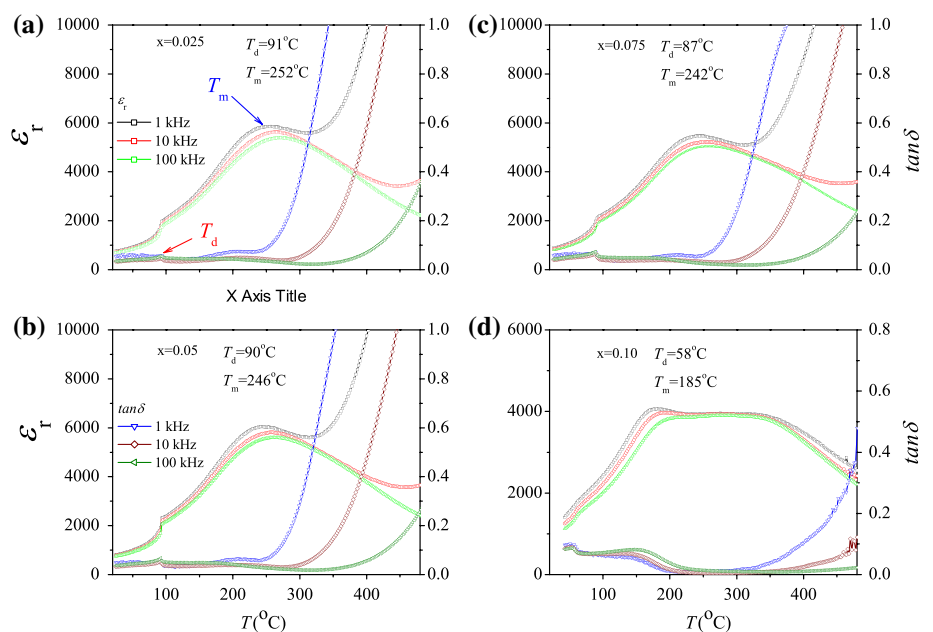
greatly with  $x$ . It can be seen that the observed  $Q_m$  decreases significantly as  $x$  increases (Fig. 4a). This suggests that after the substitution of  $\text{Li}^+$  for  $\text{Na}^+$ , the ceramics become “softened”, thus giving rise to significant improvements in  $d_{33}$ ,  $k_p$ , and  $\epsilon_r$ . As compared to the pure BNT and classic 0.94BNT–BaTiO<sub>3</sub> ceramics, the present ceramics exhibits much better piezoelectric properties. The significant improvements in piezoelectric properties of the  $[(\text{Bi}_{0.98}\text{La}_{0.02}\text{Na}_{1-x}\text{Li}_x)_{0.5}]_{0.94}\text{Ba}_{0.06}\text{TiO}_3$  ceramics should also be attributed to the lower  $E_c$ , larger  $P_r$  and the existence of MPB. According to the phenomenological theory,  $d_{33}$  is related to  $\epsilon_r$ , the spontaneous polarization  $P_s$  (which may be approximated by  $P_r$ ) and the electrostrictive coefficient  $Q_{11}$  via a general equation  $d_{33} = 2Q_{11}\epsilon_0\epsilon_r P_s$  [23]. As shown in Figs. 3 and 4, the ceramic with  $x = 0.10$  possesses a relative large  $P_r$  ( $44.9 \mu\text{C}/\text{cm}^2$ ) and a relatively high  $\epsilon_r$  (1106), and so it exhibits the largest  $d_{33}$ .

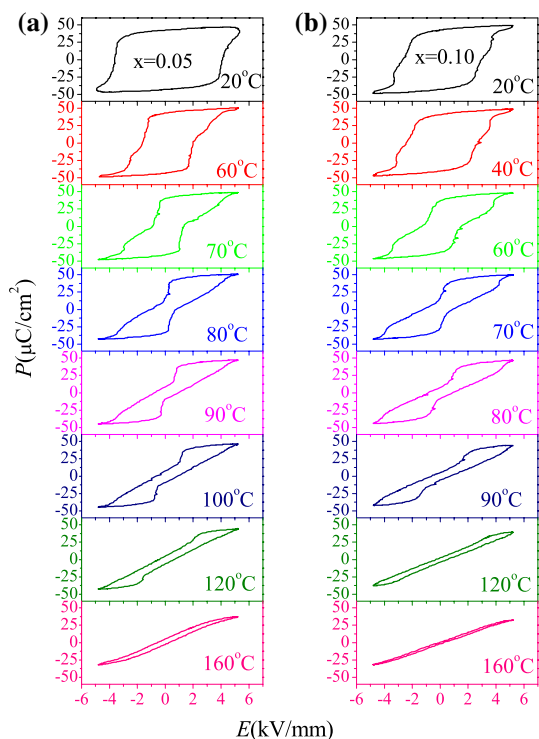
Figure 5 shows the temperature dependences of  $\epsilon_r$  and  $\tan\delta$  at 1, 10, and 100 kHz for the poled  $[(\text{Bi}_{0.98}\text{La}_{0.02}\text{Na}_{1-x}\text{Li}_x)_{0.5}]_{0.94}\text{Ba}_{0.06}\text{TiO}_3$  ceramics with  $x = 0.025, 0.05, 0.075,$  and  $0.10$ . Similar to the other BNT-based ceramics [1–10], all the  $[(\text{Bi}_{0.98}\text{La}_{0.02}\text{Na}_{1-x}\text{Li}_x)_{0.5}]_{0.94}\text{Ba}_{0.06}\text{TiO}_3$  ceramics exhibit two dielectric anomalies at  $T_d$  and  $T_m$ .  $T_d$  is the depolarization temperature which corresponds to a transition from a ferroelectric state to a so-called “anti-ferroelectric” state, while  $T_m$  is the maximum temperature at which  $\epsilon_r$  reaches a maximum value.  $T_d$  can also be derived from the peak in the temperature plot of  $\tan\delta$  [3]. As shown in Fig. 5a–d, the observed  $T_d$  for the  $[(\text{Bi}_{0.98}\text{La}_{0.02}\text{Na}_{1-x}\text{Li}_x)_{0.5}]_{0.94}\text{Ba}_{0.06}\text{TiO}_3$  ceramics decreases from 91 to 58 °C and  $T_m$  decreases from 252 to 185 °C as  $x$  increase from 0.025 to 0.10. From Fig. 5, it is also seen that for all the

ceramics,  $\epsilon_r$  exhibits a strong frequency dependence at  $T_m$ , and the maximum value of  $\epsilon_r$  decreases as frequency increases while the corresponding  $T_m$  increases, suggesting that the  $[(\text{Bi}_{0.98}\text{La}_{0.02}\text{Na}_{1-x}\text{Li}_x)_{0.5}]_{0.94}\text{Ba}_{0.06}\text{TiO}_3$  ceramics are relaxor ferroelectrics and the phase transition at  $T_m$  is a diffuse phase transition. Diffuse phase transition has been observed in many ABO<sub>3</sub>-type perovskites and bismuth layer-structured compounds, such as BNT-based ceramics [9],  $\text{K}_{0.5}\text{La}_{0.5}\text{Bi}_2\text{Nb}_2\text{O}_9$  [24],  $\text{Pb}(\text{Mg}_{1/3}\text{Nb}_{2/3})\text{O}_3$  [25], of which either the A-sites or B-sites are occupied by at least two cations. For the  $[(\text{Bi}_{0.98}\text{La}_{0.02}\text{Na}_{1-x}\text{Li}_x)_{0.5}]_{0.94}\text{Ba}_{0.06}\text{TiO}_3$  ceramics,  $\text{Na}^+$ ,  $\text{Bi}^{3+}$ ,  $\text{Li}^+$ ,  $\text{La}^{3+}$ , and  $\text{Ba}^{2+}$  are randomly distributed in the 12-fold coordination sites, so the observed diffuse phase transition behavior at  $T_m$  is reasonably attributed to the disordering of A-site cations and the local compositional fluctuation.

The variations of the  $P$ – $E$  loops with temperature for  $[(\text{Bi}_{0.98}\text{La}_{0.02}\text{Na}_{1-x}\text{Li}_x)_{0.5}]_{0.94}\text{Ba}_{0.06}\text{TiO}_3$  ceramics with  $x = 0.05$  and  $0.10$  are shown in Fig. 6. Both the ceramics exhibit a typical ferroelectric  $P$ – $E$  loop at room temperature. For the ceramic with  $x = 0.05$ , as temperature increases to 60 °C,  $E_c$  decreases and hence the loop becomes more saturated, giving a large  $P_r$  value of  $45.1 \mu\text{C}/\text{cm}^2$ . However, as temperature increases to 70 °C, the loop becomes slightly deformed but the large  $P_r$  is maintained ( $P_r = 42.7 \mu\text{C}/\text{cm}^2$ ). When temperature further increases to 80 °C, the  $P$ – $E$  loop become flatted, slanted and deformed and is different from the typical ferroelectric characteristics, and  $P_r$  decreases quickly to  $16.5 \mu\text{C}/\text{cm}^2$ . At higher temperature, the  $P$ – $E$  loop becomes slanted, narrow and slim gradually. At 160 °C, the loop becomes very slim and narrow, giving a very small value of  $P_r$

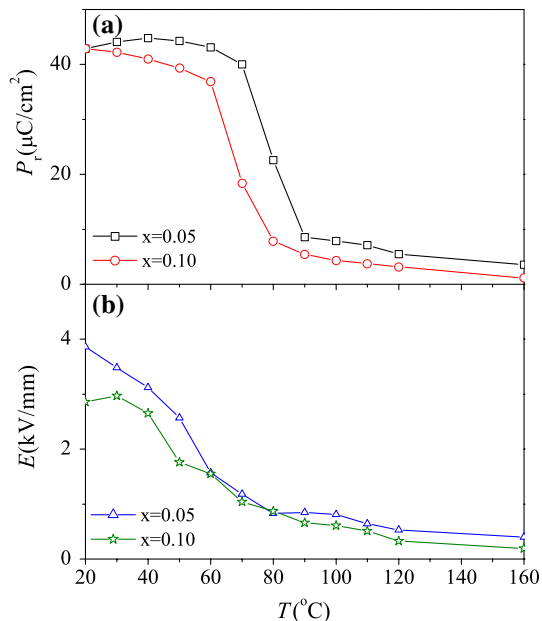
**Fig. 5** Variations of  $\epsilon_r$  and  $\tan\delta$  with temperature for the  $[(\text{Bi}_{0.98}\text{La}_{0.02}\text{Na}_{1-x}\text{Li}_x)_{0.5}]_{0.94}\text{Ba}_{0.06}\text{TiO}_3$  ceramics at 1, 10, and 100 kHz: **a**  $x = 0.025$ ; **b**  $x = 0.05$ ; **c**  $x = 0.075$ ; and **d**  $x = 0.10$





**Fig. 6**  $P$ – $E$  loops of the  $[(\text{Bi}_{0.98}\text{La}_{0.02}\text{Na}_{1-x}\text{Li}_x)_{0.5}]_{0.94}\text{Ba}_{0.06}\text{TiO}_3$  ceramics at different temperatures: **a**  $x = 0.025$ ; **b**  $x = 0.10$

( $4.89 \mu\text{C}/\text{cm}^2$ ). As determined from Fig. 6a, the depolarization temperature  $T_d$  of the ceramic with  $x = 0.05$  is about  $80$ – $90 \text{ }^\circ\text{C}$ , which is close to the value determined from the temperature plot of  $\tan\delta$  ( $91 \text{ }^\circ\text{C}$ ) (Fig. 5b). From Fig. 6b, the ceramic with  $x = 0.10$  exhibits similar temperature dependence of the ferroelectric properties, and possesses a well-saturated  $P$ – $E$  loop below  $T_d$  and a slightly deformed one above  $T_d$ , revealing a  $T_d$  value of about  $60$ – $70 \text{ }^\circ\text{C}$ . Although the loops at high temperatures ( $>T_d$ , especially  $160 \text{ }^\circ\text{C}$ ) are very slim, they are still similar to a ferroelectric hysteresis loop, and are clearly not a double-loop of antiferroelectric ceramics [26]. Besides, the forward switching field ( $E_{\text{AFE-FE}}$ ), at which the antiferroelectric domains align to become ferroelectric domains [27], cannot be observed in the loops (Fig. 6). Recently, it has been shown that, by in situ transmission electron microscopy (TEM), there is no crystallographic evidence of antiferroelectric domains near  $T_d$  [15, 27] and the depolarization is induced by the weakening of the macroscopic ferroelectric domains [27]. Together with the temperature dependence of ferroelectric and dielectric properties (Figs. 5, 6), it has been suggested that the anomalies in  $P$ – $E$  loop were resulted from the electro-mechanical interaction between the polar and non-polar regions which coexisted in the BNT-based ceramics near  $T_d$  [6, 7, 15, 28]. Therefore, on the basis of the above results, it is suggested that the polar region and non-polar region may coexist in the  $[(\text{Bi}_{0.98}\text{La}_{0.02}\text{Na}_{1-x}\text{Li}_x)_{0.5}]_{0.94}$



**Fig. 7** Variations of  $P_r$  and  $E_c$  with temperature for the  $[(\text{Bi}_{0.98}\text{La}_{0.02}\text{Na}_{1-x}\text{Li}_x)_{0.5}]_{0.94}\text{Ba}_{0.06}\text{TiO}_3$  ceramics with  $x = 0.05$  and  $0.10$

$\text{Ba}_{0.06}\text{TiO}_3$  ceramics near/above  $T_d$  and their interaction causes the  $P$ – $E$  loops become deformed at high temperatures near/above  $T_d$ .

The temperature dependences of  $P_r$  and  $E_c$  for the  $[(\text{Bi}_{0.98}\text{La}_{0.02}\text{Na}_{1-x}\text{Li}_x)_{0.5}]_{0.94}\text{Ba}_{0.06}\text{TiO}_3$  ceramics with  $x = 0.05$  and  $0.10$  are shown in Fig. 7. From Fig. 7,  $P_r$  has the large values and keeps almost unchangeable at temperature below  $T_d$ . However, as temperature increases above  $T_d$ ,  $P_r$  decreases greatly, showing clearly the ferroelectric–“non-ferroelectric” phase transition. Different from  $P_r$ ,  $E_c$  decreases with increasing temperature.

**Conclusions**

A new Li-modified BNT-based lead-free solid solution,  $[(\text{Bi}_{0.98}\text{La}_{0.02}\text{Na}_{1-x}\text{Li}_x)_{0.5}]_{0.94}\text{Ba}_{0.06}\text{TiO}_3$ , has been developed and prepared by an ordinary sintering technique. The results of X-ray diffraction reveal that the ceramics possess a pure perovskite structure. All the ceramics can be well-sintered at a relatively low sintering temperature ( $1,100^\circ\text{C}$ ). After the substitution of  $\text{Li}^+$  for  $\text{Na}^+$ , the ceramics exhibit a lower  $E_c$ , a larger  $P_r$  and thus improved piezoelectric properties. For the ceramic with  $x = 0.10$ , the piezoelectric properties become optimum, giving  $d_{33} = 212 \text{ pC/N}$  and  $k_p = 36.1\%$ . The ceramics also exhibit deformed or slim  $P$ – $E$  loops at high temperatures near/above  $T_d$ , suggesting that the ceramics may contain both the polar and non-polar regions near/above  $T_d$ .



**Acknowledgement** This work was supported by the Projects of Education Department of Sichuan Province (08ZA047), and Science and Technology Bureau of Sichuan Province (09ZQ026-059).

## References

1. Takennaka T, Maruyama K, Sakata K (1991) *Jpn J Appl Phys* 30:2236
2. Herabut A, Safari A (1997) *J Am Soc* 80:2954
3. Yoshii K, Hiruma Y, Nagata H, Takenaka T (2006) *Jpn J Appl Phys* 45:4493
4. Yu H, Ye ZG (2008) *Appl Phys Lett* 93:112902
5. Fan G, Lu W, Wang X, Liang F (2007) *Appl Phys Lett* 91:202908
6. Fan G, Lu W, Wang X, Liang F, Xiao J (2008) *J Phys D: Appl Phys* 41:035403
7. Hiruma Y, Imai Y, Watanabe Y, Nagata H, Takenaka T (2008) *Appl Phys Lett* 92:262904
8. Zhang S, Shroud TR, Nagata H, Hiruma Y, Takenaka T (2007) *IEEE Trans Ultrason Ferroelect Freq Contr* 54:910
9. Li Y, Chen W, Xu Q, Zhou J, Gu X, Fang S (2005) *Mater Chem Phys* 94:328
10. Zhang Z, Jia J, Yang H, Chen C, Sun H, Hu X, Yang D (2008) *J Mater Sci* 43:1501. doi:10.1007/s10853-007-2382-3
11. Kounga AB, Zhang ST, Jo W, Granzow T, Rödel J (2008) *Appl Phys Lett* 92:222902
12. Yao Z, Liu H, Chan L, Cao M (2009) *Mater Lett* 65:547
13. Zhou C, Liu X, Li W, Yuan C (2009) *Mater Chem Phys* 114:832
14. Zhou CR, Liu XY (2008) *J Alloys Compd* 466:563
15. Yao YQ, Tseng TY, Chou CC, Chen HHD (2007) *J Appl Phys* 102:094102
16. Yang Z, Hou Y, Liu B, Wei L (2008) *Ceram Int*. doi:10.1016/j.ceramint.2008.07.014
17. Zuo R, Ye C, Fang X, Li J (2008) *J Eur Ceram Soc* 28:871
18. Li Y, Chen W, Xu Q, Zhou J, Wang Y, Sun H (2007) *Ceram Int* 33:95
19. Chu BJ, Chen DR, Li GR, Yin QR (2002) *J Eur Ceram Soc* 22:2115
20. Zhou XY, Gu HS, Wang Y, Li WY, Zhou TS (2005) *Mater Lett* 59:1649
21. Zheng Q, Xu C, Lin D, Gao D, Kwok KW (2008) *J Phys D: Appl Phys* 41:125411
22. Zhao S, Li G, Ding A, Wang T, Yin Q (2006) *J Phys D: Appl Phys* 39:2277
23. Damjanovic D (1998) *Rep Prog Phys* 61:1267
24. Karthik C, Ravishankar N, Varma KBR (2006) *Appl Phys Lett* 89:1760
25. Uchino K, Nomura S, Cross LE, Tang SJ, Newnham RE (1980) *J Appl Phys* 51:1142
26. Haertling GH (1999) *J Am Ceram Soc* 82:797
27. Tai CWT, Choy SH, Chan HLW (2008) *J Am Ceram Soc* 91:3335
28. Suchanicz J, Kusz J, Böhm H, Duda H, Mercurio JP (2003) *J Eur Ceram Soc* 22:1559